

**Chemistry (P.G.)****[CBCS]****M.Sc. First Semester End Examination-2023****(Regular & Supplementary Paper)****PAPER-CEM-103****Full Marks: 40****Time: 02 Hrs***The figures in the right-hand margin indicate marks.**Candidates are required to give their answers in their own words as far as practicable.**Illustrate the answers wherever necessary.***Group-A**

1. Answer **any four** questions 2 × 4
- i) Differentiate between reducible and irreducible representation.
  - ii) What are the significance of globin protein in hemoglobin?
  - iii) Elaborate on Ewald sphere construction.
  - iv) Differentiate primitive and non-primitive unit cells.
  - v) Prove that if A is conjugated with B and C separately, then B and C are also conjugate with each other.
  - vi) What causes Wilson disease and Alzheimer's disease?
  - vii) What is the origin of cooperativity during oxygen binding to hemoglobin?
  - viii) Schottky defect occurs mainly in ionic lattices while Frenkel defect is predominant in covalent lattice system – explain.

iv) a) What is the significance of Hill coefficient? How J P

Collman designed 'artificial blood' in laboratory?

b) What is chelation therapy? Give a suitable example. 4+4

v) a) What do you mean by 'subgroup'? Write the conditions which must obey to form 'subgroup' Determine the sub-groups of  $D_{4h}$  and  $C_{4h}$ .

b) Verify that if there are two fold axes at right angles to one another, there must necessary be a third at right angles to both. 5 + 3

vi) a) Stoichiometric NiO is an insulator. However doping it with a small amount of  $Li_2O$  changes its electric property. Explain.

b) Give any two differences between extrinsic and intrinsic semiconductors.

c) The monoxides of the early 3d-metals are the conductors while the monoxides of the late members of the series are the semiconductors. Justify the statement

2+3+3



Group-B

2. Answer any four from the following questions 8 x 4

i) a) Name two Zn containing enzymes. Outline mechanism of action one of them.

b) Draw structures of Fe-S proteins? How they behave in presence of acid? 4+4

ii) a) Find the point group of the following molecules/ions

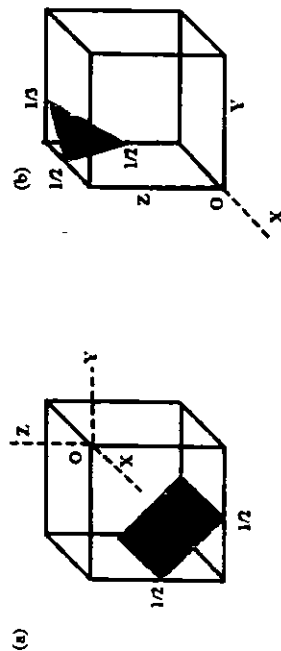
$B_3N_3H_6$ , allene, open book structure of  $H_2O_2$ , eclipse ferrocene,  $N_2O$ ,  $CO_3^{2-}$

b) Derive the matrix form of  $S_n(x)$  symmetry element. 3+5

iii) a) What are the (i) Miller indices of the planes shown in following unit cells? (ii) Determine their interplanar spacing.

Given that lattice constant,  $a = 0.25$  nm.

[Numbers shown in figure represent intercepts of the planes in terms of lattice parameters. x, y, and z represent crystallographic axes.]



b) State the properties of reciprocal lattice. How is a reciprocal lattice constructed from a direct lattice? 4 + 4