

Chemistry (P.G.)

[CBCS]

M.Sc. Third Semester End Examination-2024

(Regular & Supplementary Paper)

PAPER-CEM-302

[Inorganic Special]

Full Marks: 40

Time: 02 Hrs.

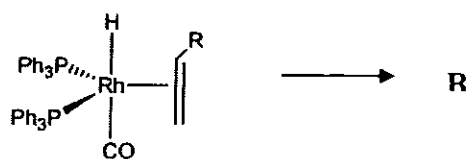
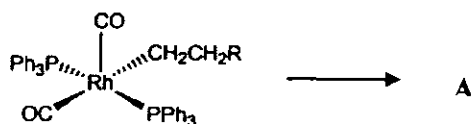
*The figures in the right hand margin indicate marks**Candidates are required to give their answers in their own words as far as practicable**Illustrate the answers wherever necessary*

Group A

Answer any four questions of the following:

4x2= 8

- (i) Differentiate between heterogeneous and homogeneous catalysts.
- (ii) What is meant by Turnover number (TON) and turnover frequency (TOF).
- (iii) Explain water gas shift reaction and mention its application.
- (iv) Identify 'A' & 'B'.



(2)

(v) What is exclusion principle?

(vi) Reduce the representation into irreducible representation.

C_{2v}	E	C_{2z}	σ_{xz}	σ_{yz}
Reducible	12	-2	2	4
Representation(Γ_{red})				

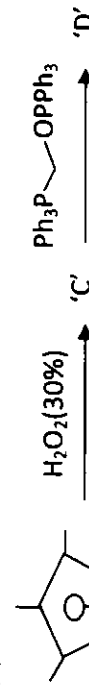
Group B

Answer any four questions of the following: $4 \times 8 = 32$

2. (i) Discuss the MO diagram of ferrocene. (4+4)
 (ii) Write two important synthesis of cyclopentadiene metal carbonyl complexes.

3. (i) Discuss the π interactions of bis arene metal complexes.

(ii) (a)



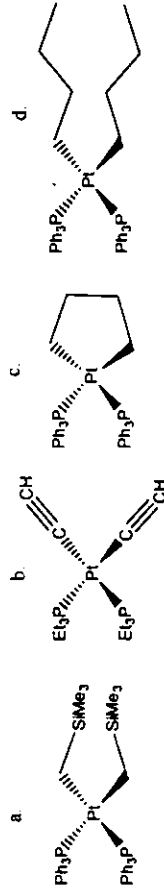
- (b) $[Cp Fe(CO)_2]_2 \xrightarrow{Na/Hg} 'E' \xrightarrow{CH_3I} 'F' + NaI$
 Identify C, D, E and F

(3)

(iii) Which catalysts are used for the hydrogenation of alkenes? (3+4+1)

4. (i) Terminal alkenes are hydrogenated more rapidly than terminal alkynes – explain.

(ii) Write down the general mechanism involved in the β -hydride elimination reaction. Write the important characteristics of this reaction.



(iii) Which of the above mentioned most stable platinum complex can undergo β -hydride elimination reaction and why? (2+3+3)

5. (i) For $[CoCl_4]^{2-}$, pure d-d transition is allowed or not? Explain.

(ii) In $[CoCl_4]^{2-}$ ion the electronic transition form $A_2 \rightarrow T_1$ is around 10 times more intense than $A_2 \rightarrow T_2$ transition. – Explain. (4+4)

(4)

6. (i) Construct the π -MO diagram of NO_2^- ion on the basis of group theory.
- (ii) Cyclobutadiene is unstable but $(\text{C}_4\text{H}_4)\text{Fe}(\text{CO})_3$ is stable-why? (4+4)

7. (i) For water molecule, verify whether the electronic transition from GS to ES is possible or not?
- (ii) Discuss the mechanism of Monsanto process. Which technique can be used to confirm the product formation and how? (5+3)

Point Group of water:

C_{2v}	E	C_{2z}	σ_{xz}	σ_{yz}
A1	1	1	1	1
A2	1	1	-1	-1
B1	1	-1	1	-1
B2	1	-1	-1	1
